



# POSTAL BOOK PACKAGE 2026

## MECHANICAL ENGINEERING

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### CONVENTIONAL Practice Sets

#### CONTENTS

#### INTERNAL COMBUSTION ENGINES

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# 1

## CHAPTER

# Introduction and Basic Concepts

## Practice Questions : Level-I

**Q.1** The brake thermal efficiency of a diesel engine is 30%. If air to fuel ratio by weight is 20 and calorific value of fuel is 41800 kJ/kg. Find brake mean effective pressure at STP (15°C and 760 mm Hg).

**Solution:**

$$\therefore \text{Brake thermal efficiency} = \frac{B.P.}{\text{Thermal power}}$$

$$0.3 = \frac{B.P.}{m_f \times C.V.} = \frac{B.P.}{\frac{m_a}{20} \times 41800}$$

$$m_a = \frac{B.P. \times 20}{0.3 \times 41800} = \frac{B.P.}{627} \text{ kg/s} \quad \dots(i)$$

Assuming volumetric efficiency 100%

$$\therefore pV_s = m_a RT$$

$$0.76 \times 13.6 \times 10^3 \times 9.81 \times V_s = \frac{B.P.}{627} \times 287 \times 288$$

$$\text{Brake mean effective pressure} = \frac{B.P.}{V_s}$$

$$= \frac{0.76 \times 13.6 \times 10^3 \times 9.81 \times 627}{287 \times 288} \text{ kPa} = 7.69 \text{ bar or } 5765.1 \text{ mm of Hg}$$

**Q2** A four cylinder 4-stroke diesel engine has a bore of 212 mm and stroke 292 mm. At full load at 720 rpm the Brake mean effective pressure is 5.93 bar and specific fuel consumption is 0.226 kg/kWH. The air fuel ratio as determined by exhaust gas analysis is 25 : 1. Calculate the brake thermal efficiency and volumetric efficiency of the engine. Atmospheric conditions are 1.01 bar and 15°C. The calorific value of fuel may be taken as 44200 kJ/kg.

**Solution:**

**Given data:**  $n = 4$ ;  $d = 212 \text{ mm}$ ;  $l = 292 \text{ mm}$ ;  $N = 720 \text{ rpm}$ ;  $p_{bm} = 5.93 \text{ bar}$   
 $\text{SFC} = 0.226 \text{ kg/kW hr.}$ ;  $A/F = 12$ ,  $C.V. = 44.2 \text{ MJ/kg}$ ;  $P = 1.01 \text{ bar}$ ,  $T_1 = 288 \text{ K}$

$$bp = \frac{p_{bm} L A N K}{60000 \times 2} = \frac{5.93 \times 10^5 \times 0.292 \times \pi / 4 \times 0.212^2 \times 4 \times \left(\frac{720}{2}\right)}{60000} = 146.69 \text{ kW}$$

$$\eta_{bth} = \frac{bp}{\dot{m}_f \times CV}$$

$$\dot{m}_f = 0.226 \times 146.69 = 33.15 \text{ kg/h}$$

$$\eta_{\text{bth}} = \frac{146.69 \times 3600}{33.15 \times 44200} = 0.36 = 36\%$$

$$\rho = \frac{p}{RT} = \frac{1.01 \times 10^5}{287 \times 288} = 1.2219 \text{ kg/m}^3$$

$$\frac{A}{F} = 25$$

$$\Rightarrow \dot{m}_a = \dot{m}_f \times 25 = 828.75 \text{ kg/h}$$

Volume flow rate at intake condition:

$$\dot{V}_a = \frac{\dot{m}_a RT}{p} = \frac{828.75}{1.2219} \times \frac{1}{60} = 11.300 \text{ m}^3/\text{min}$$

$$\dot{V}_s = \frac{\pi}{4} D^2 \frac{LNK}{2} = \frac{\pi}{4} \times 0.212^2 \times 0.292 \times 360 \times 4 = 14.84 \text{ m}^3/\text{min}$$

$$\eta_v = \frac{\dot{V}_a}{\dot{V}_s} \times 100 = \frac{11.300}{14.84} \times 100 = 76.15\%$$

**Q3** The output of an engine is given as input to an agricultural pump set. The pump is used for lifting water from a depth of 30 m at the rate of 200 litres/minute. The transmission efficiency between the engine and the pump is 100% and the pump is considered to be 100% efficient. The brake thermal efficiency of the engine is 35%, the calorific value of the fuel is 43 MJ/kg, the cost of fuel is ₹ 53.00 per litre and the density of the fuel is 780 kg/m<sup>3</sup>. Estimate the running cost of the fuel for 1000 m<sup>3</sup> of water lifted.

**Solution:**

$$\text{Power required to lift the water} = \rho g Q H = \frac{1000 \times 9.81 \times 200 \times 10^{-3} \times 30}{1000 \times 60} = 0.981 \text{ kW}$$

$$\text{Brake work done} = \text{Power} \times \text{time} = \frac{(0.981) \times (1000) \times (60)}{200 \times 10^3} = 294.3 \text{ MJ}$$

$$\text{Brake thermal efficiency} = \frac{\text{Brake work done}}{\text{Calorific value} \times \text{Mass of fuel}}$$

$$\Rightarrow 0.35 = \frac{294.3 \times 1}{43 \times \text{Mass of fuel}}$$

$$\Rightarrow \text{Mass of fuel} = 19.55 \text{ kg}$$

$$\text{Volume of fuel} = \frac{\text{Mass of fuel}}{\text{Density of fuel}} = \frac{19.55}{780} = 25.0769 \text{ litre}$$

$$\text{Running cost of engine} = 25.07 \times 53 = \text{₹ } 1328.7 \text{ only}$$

**Q4** A single cylinder 4-stroke SI engine is producing 100 kW power at an overall efficiency of 20%. Engine uses fuel-air ratio of 0.07. Determine how many m<sup>3</sup>/hr of air is used if air density is 1.2 kg/m<sup>3</sup>. The fuel vapour density is 4 times that of air. How many m<sup>3</sup>/hr of mixture is required? Calorific value of fuel is 42000 kJ/kg.

**Solution:**

$$\text{Overall efficiency, } \eta_o = \frac{BP}{\dot{m}_f \times CV}$$

$$\dot{m}_f = \frac{BP}{\eta_o \times CV} = \frac{100}{0.2 \times 42000} = 0.012 \text{ kg/s}$$

